

# **C**oastal Oceans

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# **KEY FINDINGS**

- The combustion of fossil fuels has increased carbon dioxide in the atmosphere, and the oceans have annually absorbed an equivalent of 20-30% of the carbon dioxide in fossil-fuel emissions. The present annual uptake by the oceans of approximately 1.8 billion tons of carbon (26% of global fossil-fuel emissions in 2003) is well constrained, has slightly acidified the oceans and may ultimately affect ocean ecosystems in unpredictable ways.
- The carbon budgets of ocean margins (coastal regions) are not as well-characterized due to lack of observations coupled with complexity and highly localized geographic variability. Existing data are insufficient, for example, to estimate the amount of carbon derived from human activity stored in the coastal regions of North America or to predict future scenarios.
- New air-sea carbon flux observations reveal that on average, waters within about 100 km (60 miles) of the shores surrounding North America are neither a source nor a sink of carbon dioxide to the atmosphere. A small net source of carbon dioxide to the atmosphere of 19 million tons of carbon per year (with large uncertainty) is estimated mostly from waters around the Gulf of Mexico and the Caribbean Sea. This is equivalent to about 1% of the global ocean uptake.
- With the exception of one or two time-series sites, almost nothing is known about historical trends in airsea fluxes and the source-sink behavior of North America's coastal oceans.
- The Great Lakes and estuarine systems of North America may be net sources of carbon dioxide where terrestrially-derived organic material is decomposing, while reservoir systems may be storing carbon through sediment transport and burial.
- Options for sequestering carbon in the ocean include iron fertilization in sunlit surface waters and injection of carbon dioxide in subsurface coastal waters. However, sequestration capacity and potential adverse effects on marine environments need to be investigated.
- Highly variable air-sea carbon dioxide fluxes in coastal areas may introduce errors in North American carbon dioxide fluxes calculated by atmospheric inversion methods. Reducing these errors and the uncertainties regarding the variability of carbon cycling in coastal oceans will require observation systems utilizing fixed and mobile platforms, novel instrumentation to measure critical stocks and fluxes, and coordinated national and international research programs. Experimental studies involving coastal carbon cycling should be encouraged.



Climate-driven changes in ocean circulation, chemical properties or biological rates could result in strong feedbacks to the atmosphere. This chapter first introduces the role the oceans play in modulating atmospheric

carbon dioxide (CO<sub>2</sub>), then quantifies air-sea CO<sub>2</sub> fluxes in coastal waters<sup>1</sup> surround-

ing North America and considers how the underlying processes affect the air-sea fluxes. Stocks of living organisms in marine environments are small relative to those on land, but

The uptake of this humancaused  $CO_2$  by the oceans is, on average, turning them more acidic with negative and potentially catastrophic effects on some biota. turnover rates are very high. In addition, aquatic stocks are not well characterized because of their spatial and temporal variability, the complexity of carbon compound transformations, and limited data on these processes. The oceans act as a huge reservoir

#### BOX 15.1: CCSP SAP 2.2 Uncertainty Conventions

\*\*\*\*\* = 95% certain that the actual value is within 10% of the estimate reported,
\*\*\*\* = 95% certain that the estimate is within 25%,
\*\*\* = 95% certain that the estimate is within 50%,
\*\* = 95% certain that the estimate is within 100%, and
\* = uncertainty greater than 100%.
† = The magnitude and/or range of uncertainty for the given numerical value(s) is not provided in the references cited.

for inorganic carbon, containing about 50 times as much  $CO_2$  as the atmosphere. The ocean's biological pump converts  $CO_2$  to organic particulate carbon by photosynthesis, transports the organic carbon from the surface by sinking, and therefore plays a critical role in removing atmospheric  $CO_2$  in combination with physical and chemical processes (Gruber and Sarmiento, 2002; Sarmiento and Gruber, 2006). Atmospheric concentration of  $CO_2$  would be much higher in the absence of current ocean processes implying that climate-driven changes in ocean circulation, chemical properties or biological rates could result in strong feedbacks to the atmosphere.

The release of  $CO_2$  into the atmosphere by the combustion of fossil fuels has increased pre-industrial concentrations from

Table 15.1 Climatological mean distribution of the net air-sea  $CO_2$  flux (in Gt C per year) over the global ocean regions (excluding coastal areas) in reference year 1995. The fluxes are based on about 1.75 million partial pressure measurements for  $CO_2$ in surface ocean waters, excluding the measurements made in the equatorial Pacific (10°N- 10°S) during El Niño periods (see Takahashi et *al.*, 2002). The NCAR/NCEP 42-year mean wind speeds and the (wind speed)<sup>2</sup> dependence for air-sea gas transfer rate are used (Wanninkhof, 1992). Plus signs indicate that the ocean is a source for atmospheric  $CO_2$ , and negative signs indicate that ocean is a sink. The ocean uptake has also been estimated on the basis of the following methods: temporal changes in atmospheric oxygen and  $CO_2$  concentrations (Keeling and Garcia, 2002; Bender et *al.*, 2005),  $^{13}C/^{12}C$  ratios in sea and air (Battle et *al.*, 2000; Quay et *al.*, 2003), ocean  $CO_2$ inventories (Sabine et *al.*, 2004), and coupled carbon cycle and ocean general circulation models (Sarmiento et *al.*, 2000; Gruber and Sarmiento, 2002). The consensus is that the oceans take up 1.3 to 2.3 Gt C per year.

Latitude bands	Pacific	Atlantic	Indian	Southern Ocean	Global
N of 50°N	+0.01	-0.31			-0.30
14°N-50°N	-0.49	-0.25	+0.05		-0.69
14°N-14°S	+0.65	+0.13	+0.13		+0.91
14°S-50°S	-0.39	-0.21	-0.52		-1.12
S of 50°S				-0.30	-0.30
Total flux	-0.23	-0.64	-0.34	-0.30	-1.50
% of flux	15	42	23	20	100
Area (10 <sup>6</sup> km <sup>2</sup> )	152.0	74.6	53.0	41.1	320.7
% of area	47	23	17	13	100

<sup>&</sup>lt;sup>1</sup> "Coastal waters" are the region within 100 km from shore in which processes unique to coastal marine environments influence the partial pressure of  $CO_2$  in surface sea waters.

around 280 ppm to present day levels of nearly 380 ppm in 2005. This increase in atmospheric concentrations is driving CO2 into the ocean with the present net air-sea CO<sub>2</sub> flux from the atmosphere into the ocean well constrained to about 1800 million metric tons of carbon (Mt C, See Box 15.1)\*\*\*\* per year (or 1.8 billion tons of carbon [Gt C]\*\*\*\* per year) (Figure 15.1 and Table 15.1) (See Chapter 2 for a description of how ocean carbon fluxes relate to the global carbon cycle). The uptake of this human-caused CO<sub>2</sub> by the oceans is, on average, turning them more acidic with negative and potentially catastrophic effects on some biota (Kleypas et al., 2006). The atmosphere is well mixed and nearly homogenous so the large spatial variability in air-sea CO<sub>2</sub> fluxes shown in Figure 15.1 is driven by a combination of physical, chemical, and biological processes in the ocean. The flux over the coastal margins has neither been well characterized (Liu et al., 2000) nor integrated into global calculations because there are

large variations over small spatial and temporal scales, and observations have been limited. The need for higher spatial



**Figure 15.1** Global distribution of sea-air CO<sub>2</sub> flux. The source areas (cyan-greenyellow-orange) are primarily in the tropics with a few high latitude areas where deep mixing occurs in winter. The sink areas (blue-magenta) are located in mid to high latitudes. The white line represents zero flux. Updated from Takahashi *et al.* (2002).

resolution to resolve the coastal variability has hampered modeling efforts. In the following sections we review existing information on the coastal ocean carbon cycle and its relationship to the global ocean, and we present the results of



a new analysis of about a half million observations of air-sea flux of  $CO_2$  in coastal waters surrounding the North American continent.

## 15.1.1 Global Coastal Ocean Carbon Fluxes

The carbon cycle in coastal oceans involves a series of processes, including runoff from terrestrial environments, upwelling and mixing of high  $CO_2$ water from below, photosynthesis at the sea surface, sinking of organic particles, respiration, production and consumption of dissolved organic carbon, and air-sea  $CO_2$  fluxes (Figure 15.2). Although fluxes in the coastal oceans are large relative to surface area (Muller-Karger *et al.*, 2005), there is disagreement as to whether these regions are a net sink or a net source

of  $CO_2$  to the atmosphere (Tsunogai *et al.*, 1999; Cai and Dai, 2004; Thomas *et al.*, 2004). Great uncertainties remain in coastal carbon fluxes, which are complex and dynamic, varying rapidly over short distances and at high frequencies. Only recently have new technologies allowed for the measurement of these rapidly changing fluxes (Friederich *et al.*, 1995 and 2002; Hales and Takahashi, 2004).

Carbon is transported from land to sea mostly by rivers in four components:  $CO_2$  dissolved in water, organic carbon dissolved in water, particulate inorganic carbon (*e.g.*, calcium carbonate [CaCO<sub>3</sub>]), and particulate organic carbon. The global rate of river input has been estimated to be 1000 Mt C<sup>\*\*\*</sup> per year, about 38% of it as dissolved CO<sub>2</sub> (or 384 Mt C per year), 25% as dissolved organic matter, 21% as organic particles, and 17% as CaCO<sub>3</sub> particles (Gattuso *et al.*, 1998). Estimates for the riverine dissolved CO<sub>2</sub> flux vary from 385 to 429 Mt C per year (Sarmiento and Sundquist, 1992). The Mississippi River, the seventh-largest in freshwater discharge in the world, delivers about 13 Mt C<sup>\*\*\*</sup> per year as dissolved CO<sub>2</sub> (Cai, 2003). Organic matter in continental

**Figure 15.2** Mean air-sea  $CO_2$  flux as calculated from shipboard measurements on a line perpendicular to the central California coast (top panel). Flux within Monterey Bay (~0-20 km offshore) is into the ocean, flux across the active upwelling region (~20-75 km offshore) is from the ocean, and flux in the California Current (75-300 km) is on average into the ocean. These fluxes result from the processes shown in the bottom panel. California Undercurrent water, which has a high  $CO_2$  partial pressure, upwells near shore, and is advected offshore into the California Current and into Monterey Bay. Phytoplankton growing in the upwelled water use  $CO_2$  as a carbon source, and  $CO_2$  is drawn to low levels in those areas. Phytoplankton carbon eventually sinks or is subducted below the euphotic zone, where it decays, elevating the  $CO_2$  levels of subsurface waters. Where the level of surface  $CO_2$  is higher than the level of atmospheric  $CO_2$ , diffusion drives  $CO_2$  into the atmosphere. Conversely, where the level of surface  $CO_2$  is lower than that of atmospheric  $CO_2$ , diffusion drives  $CO_2$  into the ocean. The net air-sea flux on this spatial scale is near zero. DIC = concentration of inorganic carbon (*i.e.*, all  $CO_2$  species) dissolved in seawater.; Updated from Pennington *et al.* (in press).



Table 15.2	Variability of CO <sub>2</sub> dist	ributions and fluxes in	U.S. coastal waters fro	om regional
surveys an	d moored measuremer	nts (from Doney et al.,	2004).	

Location	Surface seawater pCO <sub>2</sub> (µatm)	Instantaneous CO <sub>2</sub> flux (mol/ per m <sup>2</sup> per year)	Annual average (mol per m <sup>2</sup> per year)	Sampling method	Reference	
New Jersey Coast	211–658	-17 to +12	-0.65	Regional survey	Boehme et al. (1998)	
Cape Hatteras, North Carolina	ND	-1.0 to +1.2	ND	Moored measurements	DeGrandpre <i>et al.</i> (1 <b>997</b> )	
Middle Atlantic Bight, inner shelf	150–620	ND	-0.9	Regional survey	DeGrandpre et al. (2002)	
Middle Atlantic Bight, middle shelf	220–480	ND	-1.6	Regional survey	DeGrandpre et al. (2002)	
Middle Atlantic Bight, outer shelf	300-430	ND	-0.7	Regional survey	DeGrandpre et al. (2002)	
Florida Bay, Florida	325–725	ND	ND	Regional survey	Millero et al. (2001)	
Southern California Coastal Fronts	130–580	ND	ND	Regional survey	Simpson (1985)	
Coastal Calif. (M-1; Monterey Bay)	245–550	-8 to +50	1997–98: –1.0 1998–99: +1.1	Moored measurements	Friederich et al. (2002)	
Oregon Coast	250–640	ND	ND	Regional survey	van Geen et al. (2000)	
Bering Sea Shelf in spring (April–June)	130-400	-8 to -12	-8	Regional survey	Codispoti et al. (1986)	
South Atlantic Bight	300–1200	ND	2.5	Regional survey	Cai et al. (2003)	
Miss. River Plume (summer)	80-800	ND	ND	Regional survey	Cai et al. (2003)	
Bering Sea (Aug–Sep.)	192–400	ND	ND	Regional survey	Park et al. (1974)	

ND indicates that no data are available.

To convert from "mol" to "grams," multiply the numerical "mol" value by 12.

shelf sediments exhibits only weak isotope and chemical signatures of terrestrial origin, suggesting that riverine organic matter is reprocessed in coastal environments on a time scale of 20 to 130 years (Hedges *et al.*, 1997; Benner and Opsahl, 2001). Of the organic carbon, about 30% is accumulating in estuaries, marshes, and deltas, and a large portion (20% to 60%) of the remaining 70% is readily and rapidly oxidized in coastal waters (Smith and Hollibaugh, 1993). Only about 10% is estimated to be contributed by human activities, such as agriculture and forest clearing (Gattuso *et al.*, 1998), and the rest is a part of the natural carbon cycle.

One of the major differences between coastal and open ocean systems is the activity of the biological pump. In coastal environments, the pump operates much more efficiently, leading to rapid reduction of surface  $CO_2$  and thus complicating the accurate quantification of air-sea  $CO_2$  fluxes. For example, Ducklow and McCallister (2004) constructed a

carbon balance for the coastal oceans using the framework of the ocean carbon cycle of Gruber and Sarmiento (2002) and estimated a net  $CO_2$  removal by primary productivity of 1200 Mt C per year and a large  $CO_2$  sink of 900 Mt C per year for the atmosphere. In contrast, Smith and Hollibaugh (1993) estimated a biological pump of about 200 Mt C per year and concluded that the coastal oceans are a weak  $CO_2$ sink of 100 Mt C per year, about one-ninth of the estimate by Ducklow and McCallister (2004). Since the estimated air-sea  $CO_2$  flux depends on quantities that are not well constrained, the mass balance provides widely varying results. For this reason, in this chapter, the net air-sea flux over coastal waters is estimated on the basis of direct measurements of the air-sea difference of partial pressure of  $CO_2$  (p $CO_2$ ).

## 15.1.2 North American Coastal Carbon

Two important types of North American coastal ocean environments can be identified: (1) river-dominated coastal



**Figure 15.3** (A) Distribution of coastal surface water  $CO_2$  partial pressure measurements made between 1979 and 2004. (B) The distribution of the annual mean air-sea net  $CO_2$  flux over  $1^{\circ} \times 1^{\circ}$  pixel areas (N-S 100 km, E-W 80 km) around North America. The flux (g C per m2 per year) represents the climatological mean over the 25-year period. The magenta-blue colors indicate that the ocean water is a sink for atmospheric  $CO_2$ , and the green-yellow-orange colors indicate that the sea is a  $CO_2$  source. The data were obtained by the authors and collaborators of this chapter and are archived at the Lamont-Doherty Earth Observatory (www.ldeo.columbia.edu/res/pi/CO<sub>2</sub>).

margins with large inputs of fresh water, organic matter, and nutrients from land (*e.g.*, Mid- and South-Atlantic Bights) (Cai *et al.*, 2003) and (2) coastal upwelling zones (*e.g.*, the California-Oregon-Washington coasts, along the eastern boundary of the Pacific) where physical processes bring cool, high-nutrient, and high-CO<sub>2</sub> waters to the surface. In both environments, the biological uptake of CO<sub>2</sub> plays an important role in determining whether an area becomes a sink or a source for the atmosphere.

High biological productivity fueled by nutrients added to coastal waters can lead to seawater becoming a  $CO_2$  sink during the summer growing season, as observed in the Bering Sea Shelf (Codispoti and Friederich, 1986) and the northwest waters off Oregon and Washington (van Geen *et al.*, 2000; Hales *et al.*, 2005). Similar  $CO_2$  draw-downs may occur in the coastal waters of the Gulf of Alaska and in the Gulf of Mexico near the Mississippi River outflow. Coastal upwelling results in a very high concentration of  $CO_2$  for the surface water (as high as 1000 µatm), and, hence, the surface water becomes a strong  $CO_2$  source. This is followed by rapid biological uptake of  $CO_2$ , which causes the water to become a strong  $CO_2$  sink (Friederich *et al.*, 2002; Hales *et al.*, 2005).

A review of North American coastal carbon fluxes has been carried out by Doney *et al.* (2004) (Table 15.2). The information reviewed was very limited in space (only 13 locations) and time, leading Doney *et al.* to conclude that it was unrealistic to reliably estimate an annual flux for North American coastal waters. Measurement programs have increased recently, and we have used the newly available data to calculate annual North American coastal air-sea fluxes for the first time.

# **15.1.3 Synthesis of Available North American Air-Sea Coastal CO<sub>2</sub> Fluxes**

A large data set consisting of 550,000 measurements of the pCO<sub>2</sub> in surface waters has been assembled and analyzed (Figure 15.3; see Appendix G for details). Partial pressure of CO<sub>2</sub> is measured in a carrier gas equilibrated with sea-

water and, as such, it is a measure of the outflux/ influx tendency of  $CO_2$ from the atmosphere. Carbon dioxide reacts with seawater and 99.5% of the total amount of  $CO_2$  dissolved in sea-

The open ocean Pacific waters south of 30°N are, on the annual average, a CO<sub>2</sub> source to the atmosphere, whereas the area north of 40°N is a sink.

water is in the form of bicarbonate  $(HCO_3)$  and carbonate ions ( $CO_3^{=}$ ), which do not exchange with the overlying atmosphere. Only CO<sub>2</sub> molecules, which constitute about 0.5% of the total dissolved CO<sub>2</sub>, exchange with the atmosphere. This is expressed as  $pCO_2$ , which is affected by physical and biological processes; pCO2 increases as seawater warms and decreases when photosynthesis is stimulated. The data were obtained by the authors and collaborators, quality-controlled, and assembled in a uniform electronic format for analysis (available at www.ldeo.columbia.edu/res/pi/CO2). Observations in each  $1^{\circ} \times 1^{\circ}$  pixel area were compiled into a single year and were analyzed for time-space variability. Seasonal and interannual variations were not well characterized except in a few locations (Friederich et al., 2002). The annual mean air-sea  $pCO_2$  difference ( $\Delta pCO_2$ ) was computed for 5°-wide zones along the North American continent and was plotted as a function of latitude for four regions (Figure 15.4): North Atlantic, Gulf of Mexico/Caribbean, North Pacific, and Bering/Chukchi Seas. Figure 15.4A shows the fluxes in the first nearshore band, and Figure 15.4B shows the fluxes for a band that is several hundred kilometers from shore. The average fluxes for them and for the intermediate bands are given in Table 15.3. The flux and area data are listed in Table





Figure 15.4 Estimated air-sea CO<sub>2</sub> fluxes (g C per per m<sup>2</sup> per year) from 550,000 seawater CO<sub>2</sub> partial pressure (pCO<sub>2</sub>) observations made from 1979 to 2004 in ocean waters surrounding the North American continent. (A) Waters within one degree (about 80 km) of the coast and (B) open ocean waters between 300 and 900 km from the shore (see Figure 15.3B). The annual mean air-sea  $pCO_2$  difference ( $\Delta pCO_2$ ) values were calculated from the weekly mean atmospheric  $CO_2$  concentrations in the GLOBALVIEW-CO<sub>2</sub> database (2004) over the same pixel area in the same week and year as the seawater  $pCO_2$  was measured. The monthly net air-sea  $CO_2$  flux was computed from the mean monthly wind speeds in the National Centers for Environmental Prediction/National Center for Atmospheric Research (NCEP/ NCAR) database in the (wind speed)<sup>2</sup> formulation for the airsea gas transfer rate by Wanninkhof (1992). The ± uncertainties represent one standard deviation.

15.4. A full complement of seasonal observations are lacking in the Arctic Sea, including Hudson Bay, the northern Labrador Sea, and the Gulf of St. Lawrence; the northern Bering Sea; the Gulf of Alaska; the Gulf of California; and the Gulf of Mexico and the Caribbean Sea.

In contrast to the Pacific coast, the latitude where Atlantic coastal waters become a  $CO_2$  sink is located further north. The offshore patterns follow the same general trend found in the global open ocean data set shown in Figure 15.1. On an annual basis the lower latitudes tend to be a source of  $CO_2$  to the atmosphere, whereas the higher latitudes tend to be sinks (Figures 15.3B and 15.4B). The major difference in the coastal waters is that the latitude where  $CO_2$  starts to enter the ocean is further north than it is in the open ocean, particularly in the Atlantic. A more detailed region-by-region description follows.

### 15.1.4 Pacific Ocean

Observations made in waters along the Pacific coast of North America illustrate how widely coastal waters vary in space and time, in this case driven by upwelling and relaxation (Friederich et al., 2002). Figure 15.5A shows a summertime quasi-synoptic distribution of temperature, salinity, and pCO<sub>2</sub> in surface waters based on measurements made in July through September 2005. The effects of the Columbia River plume emanating from ~46°N are clearly seen (colder temperature, low salinity, and low pCO<sub>2</sub>), as are coastal upwelling effects off Cape Mendocino (~40°N) (colder, high salinity, and very high pCO<sub>2</sub>). These coastal features are confined to within 300 km from the coast. The 1997-2005 time-series data for surface water pCO<sub>2</sub> observed off Monterey Bay (Figure 15.5B) show the large, rapidly fluctuating air-sea CO<sub>2</sub> fluxes during the summer upwelling season in each year, as well as the low-pCO<sub>2</sub> periods during the 1997-1998 and 2002-2003 El Niño events. In spite of the large seasonal variability, ranging from 200 to 750 µatm, the annual mean air-sea pCO<sub>2</sub> difference and the net CO<sub>2</sub> flux over the waters off Monterey Bay areas (~37°N) are close to zero (Pennington et al., in press). The seasonal amplitude decreases away from the shore and in the open ocean bands, where the air-sea CO<sub>2</sub> flux changes seasonally in response to seawater temperature (out of the ocean in summer and into the ocean in winter).

The open ocean Pacific waters south of 30°N are, on the annual average, a CO<sub>2</sub> source to the atmosphere, whereas the area north of 40°N is a sink, and the zone between 30° and 40°N is neutral (Takahashi *et al.*, 2002). Coastal waters in the 40°N through 45°N zone (northern California-Oregon coasts) are even a stronger CO<sub>2</sub> sink, associated with nutrient input and stratification by fresh water from the Columbia River (Hales *et al.*, 2005). On the other hand, coastal pCO<sub>2</sub> values in the 15°N through 40°N zones have pCO<sub>2</sub> values similar to open ocean values and to the atmosphere. In the zones 15°N through 40°N, the annual mean values for the net air-sea CO<sub>2</sub> flux are nearly zero, consistent with the finding by Pennington *et al.* (in press).

#### 15.1.5 Atlantic Ocean

With the exception of the 5°N-10°N zone, the open ocean areas are an annual net sink for atmospheric CO<sub>2</sub> with stronger sinks at high latitudes, especially north of 35°N (Figure 15.3B). In contrast, the nearshore waters are a CO<sub>2</sub> source between 15°N and 45°N. Accordingly, in contrast to the Pacific coast, the latitude where Atlantic coastal waters become a CO<sub>2</sub> sink is located further north. In the areas north of 45°N, the open ocean waters are a strong CO<sub>2</sub> sink, due primarily to the cold Labrador Sea waters.

Table 15.3 Climatological mean annual air-sea  $CO_2$  flux (g C per m<sup>2</sup> per year) over the oceans surrounding North America. Negative values indicate that the ocean is a  $CO_2$  sink for the atmosphere. N is the number of seawater p $CO_2$  measurements. The ± uncertainty is given by one standard deviation of measurements used for analysis and represents primarily the seasonal variability.

Ocean regions	Coastal boxes <sup>a</sup>		First offshore <sup>a</sup>		Second offshore <sup>a</sup>		Third offshore <sup>a</sup>		Open ocean <sup>a</sup>	
	Flux	Ν	Flux	N	Flux	Ν	Flux	N	Flux	Ν
North Atlantic	3.2±142	80,417	-1.4±94	65,148	-7.3±57	35,499	−10.4± 76.4	15,771	-26±83	37,667
North Pacific	-0.2±105	164,838	-6.0±81	69,856	-4.3±66	32,045	-5.3±60	16,174	-1.2±56	84,376
G. Mexico Caribbean	9.4±24	75,496	8.4±23	61,180	11.5±17.0	8,410	13±20	1,646		
Bering/ Chukchi	28.0±110	892	-28±128	868	-44±104	3,399	-53±110	1,465	-63±130	1,848

<sup>a</sup> The pCO<sub>2</sub> data are binned into 1° latitude x 1° longitude box areas. The boxes that include shorelines are named "coastal boxes," and the 1° x 1° boxes located on the ocean side of these "coastal boxes" are called "first offshore" boxes. The next two rows of ocean side boxes are called respectively the "second offshore" and the "third offshore" boxes.

In the coastal zone very high pCO<sub>2</sub> values (up to 2600  $\mu$ atm) are observed occasionally in areas within 10 km offshore of the barrier islands (see small red dots off the coasts of Georgia and the Carolinas in Figure 15.3B). These waters which have salinities around 20 and high total CO<sub>2</sub> concentrations appear to represent outflow of estuarine/marsh waters rich in carbon (Cai *et al.*, 2003). The large contribution of fresh water that is rich in organic matter relative to the Pacific contributes to this small coastal Atlantic source. Offshore fluxes are in phase with the seasonal cycle of warming and cooling; fluxes are out of the ocean in summer and fall and are the inverse in winter and spring.

### 15.1.6 Bering and Chukchi Seas

Although measurements in these high-latitude waters are limited, the relevant data for the Bering Sea (south of 65°N) and Chukchi Sea (north of 65°N) are plotted as a function of the latitude in Figure 15.4. The values for the areas north of 55°N are for the summer months only; CO<sub>2</sub> observations are not available during winter seasons. Although data scatter widely, the coastal and open ocean waters are a strong CO<sub>2</sub> sink during the summer months due to photosynthetic drawdown of CO<sub>2</sub>. The data in the 70°-75°N zone are from the shallow shelf areas in the Chukchi Sea. These waters are a very strong CO<sub>2</sub> sink (air-sea pCO<sub>2</sub> differences ranging from -80 to -180 µatm) with little changes between the coastal and open ocean areas. The air-sea CO<sub>2</sub> flux during winter months is not known but the summer fluxes are shown in Figure 15.4 for comparison. Bates (2006) estimated a meanannual air-to-sea  $CO_2$  flux<sup>2</sup> of 39 Mt C<sup>\*\*\*</sup> per year over the Chukchi shelf using data from spring and summer of 2002 that suggested that remnant winter waters were as strong a  $CO_2$  sink as summer waters (with air-sea p $CO_2$  differences of -60 to -160 µatm).

#### 15.1.7 Gulf of Mexico and Caribbean Sea

Although observations are limited, available data suggest that these waters are a strong  $CO_2$  source (Figure 15.4 and Table 15.3). A subsurface anoxic zone has been formed in the Texas-Louisiana coast as a result of the increased addition of anthropogenic nutrients and organic carbon by the Mississippi River (*e.g.*, Lohrenz *et al.*, 1999). The carbon-nutrient cycle in the northern Gulf of Mexico is also being investigated (*e.g.*, Cai, 2003), and the studies suggest that at times those waters are locally a strong  $CO_2$  sink due to high biological production.

## **15.2 SYNTHESIS**

An analysis of half a million measurements of air-sea flux of CO<sub>2</sub> shows that the nearshore (< 100 km) coastal waters surrounding North America are a net CO<sub>2</sub> source for the atmosphere on an annual average of about 19 $\pm$ 22 Mt C per year <sup>3</sup> (Table 15.4). Most of the flux (14 $\pm$ 9 Mt C per year)<sup>3</sup> occurs in the Gulf of Mexico

An analysis of half a million measurements shows that the nearshore (< 100 km) coastal waters surrounding North America are a weak net CO<sub>2</sub> source for the atmosphere, the open oceans are a net CO<sub>2</sub> sink on an annual average.

<sup>&</sup>lt;sup>2</sup> The flux was estimated on the basis of measurements made only during the spring and summer months of 2002 at several stations located in a limited area of the Chukchi Sea. The uncertainty of  $\pm$  7 Mt C given in the original paper represents one standard deviation of

<sup>-</sup>

measured  $pCO_2$ , but does not include uncertainties in the sea-air gas transfer coefficient estimated on the basis of wind speeds and those from limited time-space coverage.

 $<sup>^3</sup>$  The specified uncertainty is  $\pm$  one standard deviation around the mean.

Table 15.4 Areas ( $km^2$ ) and mean annual air-sea CO<sub>2</sub> flux (Mt C per year) over four ocean regions surrounding North America. Since the observations in the areas north of 60°N in the Chukchi Sea were made only during the summer months, the fluxes from that area are not included. The ± uncertainty is given by one standard deviation of measurements used for analysis and represents primarily the seasonal variability.

Ocean areas (km²)				Mean air-sea CO <sub>2</sub> flux (10 <sup>12</sup> grams or Mt C per year)					
Coastal boxes	First offshore	Second offshore	Third offshore	Open ocean	Coast box	First offshore	Second offshore	Third offshore	Open ocean
North Atlantic coast (8° N to 45°N)									
625,577	651,906	581,652	572,969	3,388,500	2.7±9.5	-0.5±9.3	-4.0±4.9	-6.5±6.3	-41.5±28.1
North Pacific coast (8°N to 55°N)									
1,211,555	855,626	874,766	646,396	7,007,817	2.1±17.1	-7.0±14.1	-4.8±12.5	-3.7±5.3	-53.8±60.7
Gulf of Mexico and Caribbean Sea (8°N to 30°N)									
1,519,335	1,247,413	935,947	1,008,633		13.6±8.9	10.9±7.5	6.8±5.00	6.6±5.0	
Bering and Chukchi Seas (50°N to 70°N)									
481,872	311,243	261,974	117,704	227,609	0.8±3.1	-6.2±9.5	-5.3±7.5	-3.7±3.0	-9.8±3.7
Total ocean areas surrounding North America									
3,838,339 3,066,188 2,654,339 2,300,702 10,623,926 19.1±21.8 -2.8±20.7 -7.4±16.2 -7.3±10.1 -105.2±67.								-105.2±67.0	

and Caribbean Sea. The open oceans are a net CO<sub>2</sub> sink on an annual average (Table 15.4; Takahashi et al., 2002). The reported uncertainties reflect the time-space variability but do not reflect uncertainties due to lack of observations in some portions of the Arctic Sea, Bering Sea, Gulf of Alaska, Gulf of Mexico, or Caribbean Sea. Observations in these areas will be needed to improve estimates. If the estimate of 39 Mt C<sup>\*\*\*</sup> per year sink for the Chukchi Sea (Bates, 2006) is included, the North American coastal waters might be a small  $CO_2$  sink. These results are consistent with recent global estimates that suggest that nearshore areas receiving terrestrial organic carbon input are sources of CO<sub>2</sub> to the atmosphere and that marginal seas are sinks (Borges, 2005; Borges et al., in press). Hence, the net contribution from North American ocean margins is small and difficult to distinguish from zero. It is not clear how much of the open ocean sink results from photosynthesis driven by nutrients of coastal origin.

# **15.3 TRENDS AND DRIVERS**

The sea-to-air CO<sub>2</sub> flux from the coastal zone is small (about 1%) compared with the global ocean uptake flux, which is about 1800 Mt C per year (or 1.8 Gt C per year), and hence does not influence the global air-sea CO<sub>2</sub> budget. However, coastal waters undergo large variations in air-sea CO<sub>2</sub> flux on daily to seasonal time scales and on small spatial scales (Figure 15.5). Fluxes can change on the order of 250 g C per  $m^2$  per year or 0.7 g C per  $m^2$  per day on a day to day basis (Figure 15.5). These large fluctuations can significantly

modulate atmospheric  $CO_2$  concentrations over the adjacent continent and need to be considered when using the distribution of  $CO_2$  in calculations of continental fluxes.

Freshwater bodies have not been treated in this analysis except to note the large surface  $pCO_2$  resulting from estuaries along the east coast. The Great Lakes and rivers also represent net sources of  $CO_2$  as, in the same manner as the estuaries, organic material from the terrestrial environment is oxidized so that respiration exceeds photosynthesis. Interestingly, the effect of fresh water is opposite along the coast of the Pacific northwest, where increased stratification and iron inputs enhance photosynthetic activity (Ware and Thomson, 2005), resulting in a large sink for atmospheric  $CO_2$  (Figure 15.3). A similar process may be at work at the mouth of the Amazon (Körtzinger, 2003). This emphasizes once again the important role of biological processes in controlling the air-sea fluxes of  $CO_2$ .

The air-sea fluxes and the underlying carbon cycle processes that determine them (Figure 15.2) vary seasonally, interannually, and on longer time scales. The eastern Pacific, including the United States' west coast, is subject to changes associated with large-scale climate oscillations such as El Niño (Chavez *et al.*, 1999; Feely *et al.*, 2002; Feely *et al.*, 2006) and the Pacific Decadal Oscillation (PDO) (Chavez *et al.*, 2003; Hare and Mantua, 2000; Takahashi *et al.*, 2003). These climate patterns, and others, like the North Atlantic Oscillation (NAO), alter the oceanic CO<sub>2</sub> sink/source conditions directly through seawater temperature changes as well as ecosystem variations that occur via complex physical-



**Figure 15.5** Time-space variability of coastal waters off the west coast of North America. (A) Quasi-synoptic distribution of the temperature, salinity, and pCO2 in surface waters during July-September 2005. The Columbia River plume (~46°N) and the upwelling of deep waters off Cape Mendocino (~40°N) are clearly seen. (B) 1997-2005 time-series data for air-sea  $CO_2$  flux from a mooring off Monterey Bay, California. Seawater is a  $CO_2$  source for the atmosphere during the summer upwelling events, but biological uptake reduces levels very rapidly. These rapid fluctuations can affect atmospheric  $CO_2$  levels. For example, if  $CO_2$  from the sea is mixed into a static column, a 500 m thick planetary boundary layer over the course of one day, atmospheric  $CO_2$  concentration would change by 2.5 µatm. If the column of air is mixed vertically through the troposphere to 500 mbar, a change of about 0.5 µatm would occur. The effects would be diluted as the column of air mixes laterally. However, this demonstrates that the large fluctuations of air-sea  $CO_2$  flux observed over coastal waters could affect the concentration of  $CO_2$  significantly enough to affect estimates of air-land flux based on the inversion of atmospheric  $CO_2$  data. Air-sea  $CO_2$  flux was low during the 1997-1998 and 2002-2003 El Niño periods. The shaded areas indicate the 1997-1998 and 2002-2003 El Niño episodes. The greatest El Niño anomalies occur in the winter which is the period of lowest air-sea fluxes.

biological interactions (Hare and Mantua, 2000; Chavez *et al.*, 2003; Patra *et al.*, 2005). For example, during El Niño, upwelling of high CO<sub>2</sub> waters is dramatically reduced along central California (Figure 15.5) so that flux out of the ocean is reduced. At the same time, photosynthetic uptake of CO<sub>2</sub> is also reduced (Chavez *et al.* 2002), reducing ocean uptake. The net effect of climate variability on air-sea fluxes therefore remains uncertain and depends on the time-space integral of the processes.

## **15.4 OPTIONS FOR MANAGEMENT**

Two options for carbon sequestration have been proposed: (1) injection of  $CO_2$  in deep subsurface waters (Brewer, 2003) and (2) ocean iron fertilization (Martin, 1990). The first might be applicable in waters surrounding North America, although potential biological side effects are unresolved. The largest potential for iron fertilization resides in the high nutrient waters of the equatorial Pacific, subarctic Pacific, and

Southern Ocean. Offshore waters of coastal upwelling systems have also been considered to be iron limited. However, efficiency and capacity of sequestration remain unresolved (Bakker *et al.*, 2001; Boyd *et al.*, 2000; Coale *et al.*, 2004; Gervais *et al.*, 2002) as do environmental perturbations that could be induced by fertilization (Chisholm *et al.*, 2001).

# 15.5 RESEARCH AND DEVELOPMENT NEEDS VIS-À-VIS OPTIONS

Waters with highly variable air-sea  $CO_2$  fluxes are located primarily within 100 km of the coast (Figure 15.5). With the exception of a few areas, the available observations are grossly inadequate to resolve the high-frequency, smallspatial-scale variations. These high intensity air-sea  $CO_2$ flux events may introduce errors in continental  $CO_2$  fluxes calculated by atmospheric inversion methods. Achieving a comprehensive understanding of the carbon cycle in waters surrounding the North American continent will



require development of advanced technologies and sustained and inter-disciplinary research efforts. Both of these seem to be on the horizon with (1) the advent of ocean observatories that include novel fixed and mobile platforms together with developing instrumentation to measure critical stocks and fluxes and (2) national and international research programs that include the Integrated Ocean Observing System (IOOS) and Ocean Carbon and Climate Change (OC<sup>3</sup>). A more comprehensive understanding will require the development of a robust observing program that incorporates time se-



ries observations of air-sea and sinking-particulate carbon fluxes in the coastal and open ocean. Our present estimates suggest that the carbon that reaches the bottom over continental margins may be responsible for upwards of  $40\%^{***}$ of the carbon reaching the ocean seafloor (Muller-Karger *et al.*, 2005). Given the importance of aquatic systems to atmospheric CO<sub>2</sub> concentrations, these developing efforts must be strongly encouraged. Ocean carbon sequestration studies should also be continued.

